

Name: _____ Date: _____ Period: _____

Name the following compounds.

- | | | | |
|-----------------------|------------------------------------|-----------------------|---------------------------------------|
| 1. KCl | 2. BaF ₂ | 3. AlBr ₃ | 19. FeCl ₂ |
| 4. Rb ₂ S | 5. MgO | 6. ZnI ₂ | 20. Pb(OH) ₂ |
| 7. SrSe | 8. Cd ₃ N ₂ | 9. Ag ₂ S | 21. NH ₄ NO ₂ |
| 10. CsI | 11. CaTe | 12. SrF ₂ | 22. CuI |
| 13. Li ₃ N | 14. Rb ₃ P | 15. K ₂ Se | 23. BaHPO ₄ |
| 16. MgTe | 17. Mg ₃ N ₂ | 18. SrCl ₂ | 24. Cr(PO ₃) ₂ |

Write formulas for the following compounds.

- | | | |
|--------------------------|-----------------------|------------------------------|
| 1. sodium iodide | 2. Calcium iodide | 3. Calcium hydroxide |
| 4. Potassium nitride | 5. Silver sulfate | 6. Sodium hydrogen carbonate |
| 7. Calcium phosphate | 8. Sodium nitrate | 9. Calcium sulfide |
| 10. Zinc oxide | 11. Aluminum cyanide | 12. Potassium bromide |
| 13. Potassium carbonate | 14. Cadmium nitrate | 15. Sodium hydroxide |
| 16. Cesium sulfate | 17. Lithium selenide | 18. Ammonium chloride |
| 19. iron(II) hydroxide | 20. silver acetate | 21. rubidium arsenide |
| 22. nickel (III) sulfite | 23. strontium oxalate | 24. sodium carbonate |